

The ammonium chromium(III) alum $\text{NH}_4\text{Cr}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$

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Key indicators

Single-crystal X-ray study
 $T = 100 \text{ K}$
 Mean $\sigma(\text{S-O}) = 0.001 \text{ \AA}$
 Disorder in solvent or counterion
 R factor = 0.019
 wR factor = 0.051
 Data-to-parameter ratio = 12.3

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound, ammonium chromium(III) bis(sulfate) dodecahydrate, is composed of discrete NH_4^+ cations, $[\text{Cr}(\text{H}_2\text{O})_6]^{3+}$ cations, SO_4^{2-} anions and water molecules. The Cr and N atoms are located on special positions of site symmetry $\bar{3}$; the S atom and one of the O atoms bonded to it are located on a threefold rotation axis. The crystal packing is stabilized by several hydrogen bonds.

Received 10 May 2004

Accepted 17 May 2004

Online 5 June 2004

Comment

The ammonium chromium alum $\text{NH}_4\text{Cr}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$, (I), is composed of discrete NH_4^+ cations, $[\text{Cr}(\text{H}_2\text{O})_6]^{3+}$ cations, SO_4^{2-} anions and water molecules. The Cr and N atoms are located on special positions of site symmetry $\bar{3}$; the S atom and one of the O atoms bonded to it are located on a threefold rotation axis.

Alum (I) is isomorphous with $\text{KCr}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$ (Bacon & Gardner, 1958), $\text{CsCo}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$ (Beattie *et al.*, 1981), $\text{NaAl}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$ (Cromer *et al.*, 1967; Kay & Cromer, 1970), $\text{ND}_4\text{Al}(\text{SO}_4)_2 \cdot 12\text{D}_2\text{O}$ (Cromer & Kay, 1967), $\text{KAl}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$ (Beevers & Lipson, 1934), $\text{RbAl}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$ (Larson & Cromer, 1967), $\text{NH}_4\text{Al}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$ (Larson & Cromer, 1967; Abdeen *et al.*, 1981) and $\text{KV}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$ (Beattie *et al.*, 1996).

Although Nyburg *et al.* (2000) have discovered that the SO_4^{2-} group is disordered in many α -alums, we did not find this kind of disorder in the title compound.

Experimental

The title compound was prepared according to Gmelin (1962). Ethanol (7 ml) was dropped into a solution of 10 g $(\text{NH}_4)_2\text{Cr}_2\text{O}_7$ in 100 ml water and 11 ml H_2SO_4 . Immediately, $\text{NH}_4\text{Cr}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$ precipitated from the reaction solution. Crystals of the title compound suitable for X-ray analysis were obtained by recrystallization from water at ambient temperature.

Crystal data

$\text{NH}_4\text{Cr}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$
 $M_r = 478.35$
 Cubic, $P\bar{a}3$
 $a = 12.2491(12) \text{ \AA}$
 $V = 1837.9(3) \text{ \AA}^3$
 $Z = 4$
 $D_x = 1.729 \text{ Mg m}^{-3}$
 Mo $K\alpha$ radiation

Cell parameters from 4846 reflections
 $\theta = 3.7\text{--}27.6^\circ$
 $\mu = 0.94 \text{ mm}^{-1}$
 $T = 100(2) \text{ K}$
 Block, violet
 $0.24 \times 0.22 \times 0.18 \text{ mm}$

Data collection

Stoe IPDS-II two-circle diffractometer
 ω scans
 Absorption correction: multi-scan (MULABS; Spek, 1990; Blessing, 1995)
 $T_{\min} = 0.805$, $T_{\max} = 0.849$
 5024 measured reflections

714 independent reflections
 613 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.026$
 $\theta_{\text{max}} = 27.7^\circ$
 $h = -12 \rightarrow 15$
 $k = -14 \rightarrow 3$
 $l = -15 \rightarrow 15$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.019$
 $wR(F^2) = 0.051$
 $S = 1.04$
 714 reflections
 58 parameters
 H atoms treated by a mixture of
 independent and constrained
 refinement

$w = 1/[\sigma^2(F_o^2) + (0.0348P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.29 \text{ e } \text{Å}^{-3}$
 $\Delta\rho_{\min} = -0.36 \text{ e } \text{Å}^{-3}$
 Extinction correction: *SHELXL97*
 Extinction coefficient: 0.090 (3)

Table 1

Selected bond distances (Å).

Cr1—O3 ⁱ	1.9602 (9)	S1—O2	1.4709 (18)
S1—O1	1.4704 (9)		

Symmetry code: (i) y, z, x .

Table 2

Hydrogen-bonding geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
N1—H1 ⁱ ···O4	0.85 (3)	2.16 (3)	3.0049 (11)	173 (3)
O3—H3A···O1 ^{viii}	0.80 (2)	1.84 (2)	2.6293 (13)	172 (2)
O3—H3B···O4 ^{ix}	0.83 (2)	1.78 (2)	2.6071 (13)	173 (2)
O4—H4B···O2 ^x	0.79 (3)	1.99 (3)	2.7752 (15)	171 (2)
O4—H4A···O1 ^{xi}	0.74 (2)	2.02 (2)	2.7454 (14)	169 (2)

Symmetry codes: (viii) $z, \frac{3}{2} - x, y - \frac{1}{2}$; (ix) $\frac{3}{2} - x, 1 - y, z - \frac{1}{2}$; (x) $\frac{1}{2} + x, \frac{3}{2} - y, 1 - z$; (xi) $\frac{1}{2} + z, x, \frac{3}{2} - y$.

H atoms bonded to O atoms were refined isotropically. H atoms bonded to N atoms are disordered. Their coordinates were refined with fixed individual displacement parameters [$U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{N})$].

Data collection: *X-AREA* (Stoe & Cie, 2001); cell refinement: *X-AREA*; data reduction: *X-AREA*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *XP* in *SHELXTL-Plus* (Sheldrick, 1991); software used to prepare material for publication: *SHELXL97* and *PLATON* (Spek, 1990).

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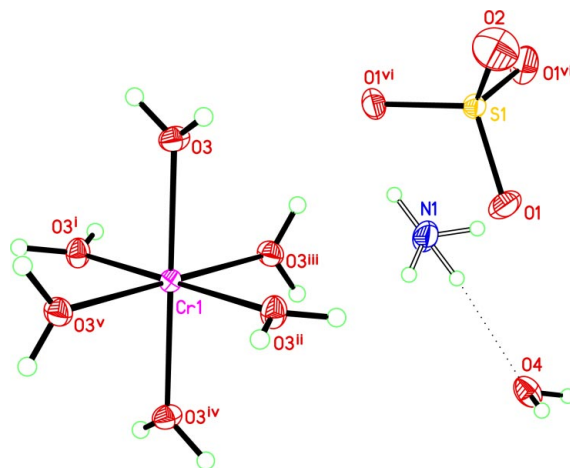


Figure 1

Perspective view of the title compound with the atom numbering; displacement ellipsoids are at the 50% probability level. Only four of the six disordered H atoms of the ammonium group are shown. [Symmetry codes: (i) y, z, x ; (ii) $-y, -z, -x$; (iii) $1 - z, 1 - x, 1 - y$; (iv) $1 - x, 1 - y, 1 - z$; (v) z, x, y ; (vi) $\frac{3}{2} - y, 1 - z, x - \frac{1}{2}$; (vii) $\frac{1}{2} + z, \frac{3}{2} - x, 1 - y$.]

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